

## Note

Partial benzoylation of  $\beta$ -gentiobiose

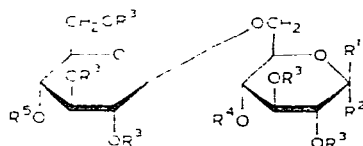
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The selective acylations of a series of (1 $\rightarrow$ 4)-D-linked oligosaccharides<sup>1-6</sup> and their derivatives<sup>6-10</sup> have been previously investigated. No report has been found, however, on the preferential acylation of oligosaccharides having other glycosidic linkages. This paper describes the partial benzoylation of  $\beta$ -gentiobiose (6-*O*- $\beta$ -D-glucopyranosyl- $\beta$ -D-glucopyranose)<sup>11</sup> (**1**) with benzoyl chloride in pyridine.

Reaction of **1** with 8 mol. equiv. of benzoyl chloride in pyridine at  $-40^\circ$  gave a mixture of four products (t.l.c.) which was fractionated by column chromatography. The first-eluted component was obtained in crystalline form in 31% yield and identified as  $\beta$ -gentiobiose octabenzoate (**2**) by comparison with an authentic specimen synthesized from **1** with an excess of benzoyl chloride.



- |  |   |
|--|---|
| 1 $R = \text{OH}, R^2 = R^3 = R^4 = R^5 = \text{H}$                          | 8 $R^1 = \text{OBz}, R^2 = R^4 = \text{H}, R^3 = R^5 = \text{Bz}$             |
| 2 $R^1 = \text{OBz}, R^2 = \text{H}, R^3 = R^4 = R^5 = \text{Bz}$            | 9 $R^1 = \text{OBz}, R^2 = \text{H}, R^3 = R^5 = \text{Bz}, R^4 = \text{Me}$  |
| 3 $R^1 = \text{OBz}, R^2 = R^5 = \text{H}, R^3 = R^4 = \text{Bz}$            | 10 $R^1 = \text{OBz}, R^2 = \text{H}, R^3 = R^5 = \text{Bz}, R^4 = \text{Ac}$ |
| 4 $R^1 = R^5 = \text{H}, R^2 = \text{CBz}, R^3 = R^4 = \text{Bz}$            | 11 $R^1 = \text{CBz}, R^2 = R^4 = R^5 = \text{H}, R^3 = \text{Bz}$            |
| 5 $R^1 = \text{CBz}, R^2 = \text{H}, R^3 = R^4 = \text{Bz}, R^5 = \text{Me}$ | 12 $R^1 = \text{CBz}, R^2 = \text{H}, R^3 = \text{Bz}, R^4 = R^5 = \text{Me}$ |
| 6 $R^1 = \text{H}, R^2 = \text{CBz}, R^3 = R^4 = \text{Bz}, R^5 = \text{Me}$ | 13 $R^1 = \text{CBz}, R^2 = \text{H}, R^3 = \text{Bz}, R^4 = R^5 = \text{Ac}$ |
| 7 $R^1 = \text{OAc}, R^2 = \text{H}, R^3 = R^4 = \text{Ac}, R^5 = \text{Me}$ |   |

The second component eluted from the column moved as a single component on t.l.c. in a variety of solvent systems, and was obtained as an amorphous powder in 26% yield. Elementary analysis showed that it was a hepta-*O*-benzoyl derivative which subsequently was shown to be a 2:1 mixture of 1,2,3,4,2',3',6'-hepta-*O*-benzoyl- $\beta$ - (**3**) and  $\alpha$ -gentiobiose (**4**) as follows: The n.m.r. spectrum of the mixture in chloroform-*d* solution exhibited two doublets due to the anomeric protons at  $\tau$  3.31 and 3.82 with relative proton intensities of 1:2, thus indicating the presence of a 2:1

mixture of the  $\beta$  and  $\alpha$  anomers (**3** and **4**). To determine the position of the free hydroxyl group, the mixture of **3** and **4** was converted into a mixture of the hepta-*O*-benzoyl-mono-*O*-methyl derivatives **5** and **6** by methylation with diazomethane-boron trifluoride etherate<sup>12</sup>. *O*-Debenzoylation of the mixture of **5** and **6**, followed by methanolysis, gave methyl 4-*O*-methyl- $\alpha,\beta$ -D-glucopyranoside and methyl  $\alpha,\beta$ -D-glucopyranoside, identified by g.l.c. as the *O*-trimethylsilyl derivatives, which indicated that the free hydroxyl group of **3** and **4** was located at C-4 or -4'. *O*-Debenzoylation of the mixture of **5** and **6** and subsequent acetylation gave the known 1,2,3,4,2',3',6'-hepta-*O*-acetyl-4'-*O*-methyl- $\beta$ -gentiobiose<sup>13</sup> (**7**), thus establishing the structure of **3** and **4**.

The third component eluted was obtained in crystalline form in 26% yield. Elementary analysis indicated a heptabenzoyl derivative which proved to be 1,2,3,2',3',4',6'-hepta-*O*-benzoyl- $\beta$ -gentiobiose (**8**) as follows: On methylation, **8** gave the hepta-*O*-benzoyl-mono-*O*-methyl derivative **9** which was *O*-debenzoylated and methanolized to afford methyl 4-*O*-methyl- $\alpha,\beta$ -D-glucopyranoside and methyl  $\alpha,\beta$ -D-glucopyranoside (g.l.c.). This result, combined with the data for the structural elucidation of the mixture of **3** and **4**, indicated that **8** is the positional isomer of the mixture of **3** and **4** having HO-4' free. Hence, the free hydroxyl group in **8** was assigned to HO-4. In the n.m.r. spectrum of **8** in chloroform-*d*, the anomeric proton signal appeared at  $\tau$  3.90 as a doublet with *J* 8.0 Hz, consistent with the  $\beta$  configuration at C-1. Acetylation of **8** gave the crystalline 4-*O*-acetyl-1,2,3,2',3',4',6'-hepta-*O*-benzoyl derivative **10**.

The fourth component eluted was obtained as an amorphous powder in 12% yield. The structure of 1,2,3,2',3',6'-hexa-*O*-benzoyl- $\beta$ -gentiobiose (**11**) was assigned to this compound on the basis of the results of the elementary analysis, of n.m.r. data, and sequential methylation. This gave the hexa-*O*-benzoyl-di-*O*-methyl derivative **12** which was *O*-debenzoylated and methanolized to give methyl 4-*O*-methyl- $\alpha,\beta$ -D-glucopyranoside as the sole product (g.l.c.). Acetylation of **11** afforded the 4,4'-di-*O*-acetyl-1,2,3,2',3',6'-hexa-*O*-benzoyl derivative **13** in crystalline form.

On the basis of the yields of the reaction products, HO-4 and -4' of **1** have the lowest, similar reactivities. The remarkable inertness of HO-3 towards benzoylation with benzoyl chloride in pyridine in maltose<sup>2</sup>, methyl  $\beta$ -lactoside<sup>7</sup>, methyl  $\beta$ -cellobioside<sup>6</sup>, and  $\beta$ -cyclodextrin<sup>5</sup> is known. It was also shown<sup>6</sup> that HO-3 in  $\beta$ -maltose monohydrate and methyl  $\beta$ -maltoside monohydrate is the least reactive group towards benzoylation, followed by HO-4'. The very slow benzoylation of HO-3 in maltose has been attributed<sup>2</sup> to the high steric hindrance due to the (1 $\rightarrow$ 4)- $\alpha$ -D-glycosidic linkage, which causes the close proximity of the two D-glucopyranose rings and a strong intramolecular hydrogen bond between HO-3 and HO-2'. In **1**, which has a (1 $\rightarrow$ 6)-D- $\beta$ -glycosidic linkage, no or at least only little steric interactions exist between the two monosaccharide moieties. The lack of the steric hindrance in **1** may be responsible for the observed reactivity of HO-4 and 4', lower than that of HO-3, whereas the inverse order was observed for  $\beta$ -maltose monohydrate and methyl  $\beta$ -maltoside monohydrate<sup>6</sup>.

## EXPERIMENTAL

*General methods.* — Unless otherwise stated, the general experimental conditions were the same as those described previously<sup>6</sup>. G.l.c. was performed under the same conditions as described previously<sup>14</sup>. N.m.r. spectra were recorded on chloroform-*d* solutions with tetramethylsilane as the internal standard. T.l.c. and column chromatography were performed with 9:1 (v/v) benzene-ethyl acetate as eluent.

*1,2,3,4-Tetra-O-benzoyl-6-O-(2,3,4,6-tetra-O-benzoyl-β-D-glucopyranosyl)-β-D-glucopyranose (2).* — To a cooled solution of **1** (200 mg) in dry pyridine (8 ml) was added benzoyl chloride (0.8 ml), and the mixture was kept for 2 days at room temperature, and then poured into ice-water. The precipitate formed was filtered off, washed with water, and dried. Crystallization from ethanol-acetone gave **2** (588 mg, 87%), m.p. 189–191°,  $[\alpha]_D^{20} + 22.8^\circ$  (*c* 1.6, chloroform); n.m.r.  $\tau$  3.80 (d, 1 H,  $J_{1,2}$  7.5 Hz, H-1).

*Anal.* Calc. for  $C_{68}H_{54}O_{19}$ : C, 69.50; H, 4.63. Found: C, 69.72; H, 4.68.

*Benzoylation of 1 with 8 mol. equiv. of benzoyl chloride.* — Benzoyl chloride (10.75 ml, 8 mol. equiv.) was added dropwise over a period of 20 min to a stirred solution of **1** (3.95 g) in anhydrous pyridine (160 ml) at  $-40^\circ$ . The reaction mixture was stirred for 1 h at  $-30^\circ$ , 2 h at  $-20^\circ$ , and 1 h at  $0^\circ$ , and then poured into ice-water. The resulting precipitate was filtered off, washed extensively with water, and dried. T.l.c. showed the presence of four spots having  $R_F$  values of 0.82 (**2**), 0.65 (**3** and **4**), 0.52 (**8**), and 0.25 (**11**), respectively. The mixture was fractionated on a column of Silica gel (400 g). The first fraction gave **2** (4.16 g, 30.7%), m.p. and mixed m.p. 189–191° (ethanol-acetone),  $[\alpha]_D^{17} + 22.0^\circ$  (*c* 1.0, chloroform).

The second fraction gave a 2:1 mixture of 1,2,3,4-tetra-*O*-benzoyl-6-*O*-(2,3,6-tri-*O*-benzoyl-β-D-glucopyranosyl)-β- (**3**) and α-D-glucopyranose (**4**) (3.21 g, 26.0%) as an amorphous powder,  $[\alpha]_D^{17} + 54.8^\circ$  (*c* 1.5, chloroform); n.m.r.  $\tau$  3.31 (d,  $J_{1,2}$  3.0 Hz, H-1 of **4**) and 3.82 (d,  $J_{1,2}$  8.0 Hz, H-1 of **3**) (ratio of peaks at  $\tau$  3.31 and 3.82, 1:2).

*Anal.* Calc. for  $C_{61}H_{50}O_{18}$ : C, 68.41; H, 4.71. Found: C, 68.30; H, 4.78.

The third fraction gave 1,2,3-tri-*O*-benzoyl-6-*O*-(2,3,4,6-tetra-*O*-benzoyl-β-D-glucopyranosyl)-β-D-glucopyranose (**8**) (3.22 g, 26.1%), m.p. 223–225° (methanol-acetone),  $[\alpha]_D^{17} + 9.4^\circ$  (*c* 1.7, chloroform); n.m.r.  $\tau$  3.90 (d, 1 H,  $J_{1,2}$  8.0 Hz, H-1).

*Anal.* Calc. for  $C_{61}H_{50}O_{18}$ : C, 68.41; H, 4.71. Found: C, 68.36; H, 4.62.

The fourth fraction gave 1,2,3-tri-*O*-benzoyl-6-*O*-(2,3,6-tri-*O*-benzoyl-β-D-glucopyranosyl)-β-D-glucopyranose (**11**) (1.33 g, 11.9%) as an amorphous powder,  $[\alpha]_D^{20} + 22.0^\circ$  (*c* 1.7, chloroform); n.m.r.  $\tau$  3.90 (d, 1 H,  $J_{1,2}$  8.0 Hz, H-1).

*Anal.* Calc. for  $C_{54}H_{46}O_{17}$ : C, 67.08; H, 4.80. Found: C, 67.20; H, 4.74.

*Mixture of 1,2,3,4-tetra-O-benzoyl-6-O-(2,3,6-tri-O-benzoyl-4-O-methyl-β-D-glucopyranosyl)-β- (5) and -α-D-glucopyranose (6).* — Diazomethane in dichloromethane was gradually added to a cooled solution of the mixture of **3** and **4** (1.1 g) in dichloromethane (8 ml) containing  $BF_3$  etherate (0.1 ml) until a pale-yellow color persisted, and the mixture was kept for 2 h at room temperature. Polymethylene was

filtered off, and the filtrate was washed successively with aqueous  $\text{NaHCO}_3$  and water, dried ( $\text{Na}_2\text{SO}_4$ ), and evaporated. The residue was fractionated on a column of Silica gel (30 g), and the fraction containing the major product was evaporated to give a 2:1 mixture of **5** and **6** as an amorphous solid (940 mg, 87%),  $[\alpha]_D^{20} + 54.6^\circ$  (*c* 1.7, chloroform); n.m.r.  $\tau$  3.27 (d,  $J_{1,2}$  3.5 Hz, H-1 of **6**), 3.82 (d,  $J_{1,2}$  8.0 Hz, H-1 of **5**) (ratio of peaks at  $\tau$  3.27 and 3.82, 1:2), and 6.64 (s, 3 H, OMe).

*Anal.* Calc. for  $\text{C}_{62}\text{H}_{52}\text{O}_{18}$ : C, 68.63; H, 4.83. Found: C, 68.80; H, 4.76.

A solution of the mixture of **5** and **6** (100 mg) in dry methanol (5 ml) was treated with methanolic *m* sodium methoxide (0.1 ml). The solution was kept for 3 h at room temperature, and then neutralized with Amberlite IR-120 ( $\text{H}^+$ ) ion-exchange resin and evaporated to dryness. Methanolysis of the residue (1% methanolic HCl, 5 ml; reflux, 16 h) and g.l.c. of the resulting methyl glycosides as their per-*O*-(trimethylsilyl) derivatives gave peaks corresponding to methyl 4-*O*-methyl- $\alpha,\beta$ -D-glucopyranoside (8.4 and 9.6 min) and methyl  $\alpha,\beta$ -D-glucopyranoside (13.8 and 15.2 min).

*1,2,3,4-Tetra-O-acetyl-6-O-(2,3,6-tri-O-acetyl-4-O-methyl- $\beta$ -D-glucopyranosyl)- $\beta$ -D-glucopyranose (7).* — The mixture of **5** and **6** (300 mg) was *O*-debenzoylated, as just described, and the residue was acetylated with 1:1 (v/v) acetic anhydride-pyridine (1 ml) overnight at room temperature. Isolation in the usual way gave **7** (155 mg, 81%), m.p. 158–159° (ethanol),  $[\alpha]_D^{20} - 10.8^\circ$  (*c* 0.9, chloroform); lit.<sup>13</sup>: m.p. 159–160°,  $[\alpha]_D^{25} - 10.2^\circ$ .

*1,2,3-Tri-O-benzoyl-4-O-methyl-6-O-(2,3,4,6-tetra-O-benzoyl- $\beta$ -D-glucopyranosyl)- $\beta$ -D-glucopyranose (9).* — Treatment of **8** (1.21 g) with diazomethane- $\text{BF}_3$  etherate in dichloromethane, as just described, afforded **9** (1.06 g, 86%), m.p. 176–177° (ethanol-chloroform),  $[\alpha]_D^{20} + 5.5^\circ$  (*c* 1.9, chloroform); n.m.r.  $\tau$  3.94 (d, 1 H,  $J_{1,2}$  8.0 Hz, H-1) and 6.84 (s, 3 H, OMe).

*Anal.* Calc. for  $\text{C}_{62}\text{H}_{52}\text{O}_{18}$ : C, 68.63; H, 4.83. Found: C, 68.75; H, 4.76.

*O*-Debenzoylation of **9** (100 mg), followed by methanolysis as just described, and g.l.c. of the methanolizate as the per(trimethylsilyl) ethers gave peaks corresponding to methyl 4-*O*-methyl- $\alpha,\beta$ -D-glucopyranoside and methyl  $\alpha,\beta$ -D-glucopyranoside.

*1,2,3-Tri-O-benzoyl-4-O-methyl-6-O-(2,3,6-tri-O-benzoyl-4-O-methyl- $\beta$ -D-glucopyranosyl)- $\beta$ -D-glucopyranose (12).* — Methylation of **11** (920 mg), as just described, gave **12** (840 mg, 89%), m.p. 205–206° (ethanol),  $[\alpha]_D^{18} + 13.2^\circ$  (*c* 1.4, chloroform); n.m.r.  $\tau$  3.97 (d, 1 H,  $J_{1,2}$  8.0 Hz, H-1), 6.60, and 6.85 (s, 6 H, 2 OMe).

*Anal.* Calc. for  $\text{C}_{56}\text{H}_{50}\text{O}_{17}$ : C, 67.60; H, 5.07. Found: C, 67.48; H, 5.13.

After sequential *O*-debenzoylation of **12**, methanolysis, and trimethylsilylation, g.l.c. examination showed the presence of methyl 4-*O*-methyl- $\alpha,\beta$ -D-glucopyranoside as the sole product.

*4-O-Acetyl-1,2,3-tri-O-benzoyl-6-O-(2,3,4,6-tetra-O-benzoyl- $\beta$ -D-glucopyranosyl)- $\beta$ -D-glucopyranose (10) and 4-O-acetyl-6-O-(4-O-acetyl-2,3,6-tri-O-benzoyl- $\beta$ -D-glucopyranosyl)-1,2,3-tri-O-benzoyl- $\beta$ -D-glucopyranose (13).* — Conventional acetylation of **8** (150 mg) with 1:1 (v/v) acetic anhydride-pyridine (3 ml) gave **10** (134 mg, 86%),

m.p. 191–192° (from ethanol–chloroform),  $[\alpha]_D^{20} + 52.2^\circ$  (*c* 1.6, chloroform); n.m.r.  $\tau$  3.89 (d, 3 H,  $J_{1,2}$  7.5 Hz, H-1) and 8.12 (s, 3 H, OAc).

*Anal.* Calc. for  $C_{63}H_{52}O_{19}$ : C, 67.98; H, 4.71. Found: C, 67.80; H, 4.88.

Acetylation of **11** (150 mg) afforded **13** (145 mg, 89%), m.p. 218–219° (ethanol–chloroform); n.m.r.  $\tau$  3.81 (d, 1 H,  $J_{1,2}$  8.0 Hz, H-1), 8.08, and 8.13 (s, 6 H, 2 OAc).

*Anal.* Calc. for  $C_{58}H_{50}O_{19}$ : C, 66.28; H, 4.80. Found: C, 66.20; H, 4.91.

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